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describe a chemical reaction using
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Modern Chemistry Chapter 8. A number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. When

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the positive ion of one compound replaces the positive ion of another to form two new compounds.

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organized according to the ease with
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chemical reactions. chemical equation.
represents, with symbols and formulas, the
identities and relative amounts of the
reactants and products in a chemical
reaction.

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CHAPTER 8 REVIEW Class SËc7îB-n
Chemical Equations and Reactions

SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied

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to the reactants.

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Chapter 8 (Modern Chemistry) Chemical Reactions; 2 Section 11.1 Describing Chemical Reactions. OBJECTIVES ; Describe how to write a word equation. 3 Section 11.1 Describing Chemical Reactions. OBJECTIVES ; Describe how to write a skeleton equation. 4 Section 11.1 Describing Chemical Reactions.

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OBJECTIVES

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CHAPTER 8 REVIEW Chemical

Equations and Reactions SECTION 2

SHORT ANSWER Answer the following questions in the space provided. 1. Match

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the equation type on the left to its representation on the right. c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX$ a double-displacement (d) $AX \rightarrow A + X$

8 Chemical Equations and Reactions

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