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The results are:  $\rho(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$  for the unit of kg/m<sup>3</sup>, (or,  $\rho(z) = (1.20252 - 0.101674z + 0.0022375z^2) \times 10^9$  for the unit of kg/km<sup>3</sup>) where z is the vertical distance from the earth surface at sea level. At z = 7 km, the equation would give  $\rho = 0.60$  kg/m<sup>3</sup>.

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Pg = 80 kPa A = 35 cm<sup>2</sup> Patm = 95 kPa mP = 4 kg 60 N Cengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 56 © The McGraw\u2012Hill Companies, 2008 pressure is 100 kPa.

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State 1 State 2 Process path Property B Property A FIGURE 2 \u2013 Yunus Cengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 33 © The McGraw\u2012Hill Companies, 2008 process completely, one should specify the initial and final states of the process, as well as the path it follows, and the interactions with the surroundings.

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The efficiency of a cook- ing appliance is defined as the ratio of the useful energy transferred to the hoverall \ufffd h combustion h thermal generator \ufffd W # net, electric HHV m# net 86 | Introduction to Thermodynamics and Heat Transfer TABLE 3 \u2013 Yunus Cengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 33 © The McGraw\u2012Hill Companies, 2008 The efficacy of different lighting systems Efficacy, Type of lighting lumens/W Combustion Candle 0.2 Incandescent Ordinary 6 \u2013 Yunus Cengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 33 © The McGraw\u2012Hill Companies, 2008

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