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The Elements by

Theodore Gray

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#5 How to find the

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parents and
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Atoms, elements

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does it have?

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down) columns in
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towers. periods.
atomic numbers
<p>groups</p> ...

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document to help students review a chemistry unit on atomic structure and the periodic table. I typically hand out this document to students at the end of the unit to clarify what content will be covered on an upcoming unit test

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travel through history as we examine how atoms were discovered from Democritus through Dalton, JJ Thomas, Rutherford and Bohr. We'll learn about the layout of the periodic table and discuss ions, isotopes and

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3. Atoms & the
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Mrs. Sumerix

Atoms have a mass that is based largely on the number of protons and neutrons in their nucleus. 2.6: Arrangements of Electrons Electrons are organized into

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Atoms And

Shells and Table

subshells about the
nucleus of an
atom. 2.7: The

Periodic Table The
chemical elements
are arranged in a
chart called the
periodic table.

2: Elements,
Atoms, and the
Periodic Table -
Chemistry ...

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Atoms And

Periodic Table

Directions: Label the trends of the periodic table by adding arrows and descriptions.

&

Page 2 of 2 9.1 Use

a periodic table of

the elements to

answer these

questions. 1. The

following graphics

represent the

nuclei of atoms.

Using a periodic

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Periodic Table,
fill in the table. 2.
Review Answer
Key
Look at a periodic
table. The atomic
mass of hydrogen
is 1.00794.

Periodic Table

Review

The atomic masses
now appear in the
Periodic Table is
the average atomic
mass weighted by

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Atoms And

the abundance of each isotope. Unfortunately, the abundance depends on location where the sample is taken (ultimately depends on the process that created, transported or aggregated the material).

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Pt-195 is closest to the weighted average (found on periodic table) and would be the most abundant. If an ion charge is not given, locate the electrons of the element by looking to the atomic

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number. Potassium is the only atom of these three that would have 19 electrons. Zn 30 67, Ra 88 226, Al 3 + 13 27, Mg 12 23 and P 3 – 15 31

3.E: Atoms and the
Periodic Table
(Exercises) -
Chemistry ...
The elements can

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Atoms And

be placed in the periodic table. The position of an element provides information about its properties. Most elements are metals, with different properties to those of non-metals.

The periodic table -

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TEACHER. Terms in

this set (22) atom.

the smallest unit of

an element that

maintains the

properties of that

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element. electron.
a subatomic
particle that has a
negative charge.
nucleus.

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1) Every element is
made of tiny,
unique particles
called atoms that
cannot that cannot

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Atoms And

be subdivided. 2)
Atoms of the same
element are
exactly alike. 3)
Atoms of different
elements can join
to form molecules.

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Welcome! I am Nurse Cheung. Here, I discuss all things pre-nursing

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